

This article was downloaded by:

On: 30 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

### Synthesis and investigation of new types of dicoordinated phosphorus compounds

L. N. Markovski<sup>a</sup>; V. D. Romanenko<sup>a</sup>; A. V. Kirsanov<sup>a</sup>

<sup>a</sup> Institute of Organic Chemistry Academy of Sciences of the Ukrainian SSR, Kiev, U.S.S.R.

**To cite this Article** Markovski, L. N. , Romanenko, V. D. and Kirsanov, A. V.(1983) 'Synthesis and investigation of new types of dicoordinated phosphorus compounds', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 18: 1, 31 — 34

**To link to this Article:** DOI: 10.1080/03086648308075960

**URL:** <http://dx.doi.org/10.1080/03086648308075960>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## SYNTHESIS AND INVESTIGATION OF NEW TYPES OF DICOORDINATED PHOSPHORUS COMPOUNDS

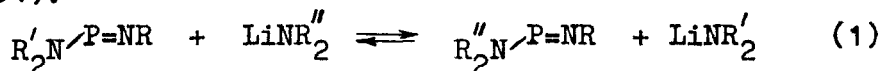
L.N.MARKOVSKI, V.D.ROMANENKO AND A.V.KIRSANOV  
Institute of Organic Chemistry Academy of Sciences  
of the Ukrainian SSR, 252094, Kiev, U.S.S.R.

Abstract New synthetic methods for the preparation of stable compounds with P=N, P=C and P=P -linkages have been developed.

### $\lambda^3$ -IMINOPHOSPHINES

Thermal elimination of small, energetically stable molecules  $XHlg$  ( $X=Me_3Si, Li$ ) from sterically crowded phosphorus-nitrogen dyad  $-P(Hlg)-N(X)-$  has been so far the only synthetic access to  $\lambda^3$ -iminophosphines.<sup>1</sup> We have elaborated a novel route to iminophosphines via nucleophilic displacement at the dicoordinated phosphorus atom in the aminoiminophosphine series.

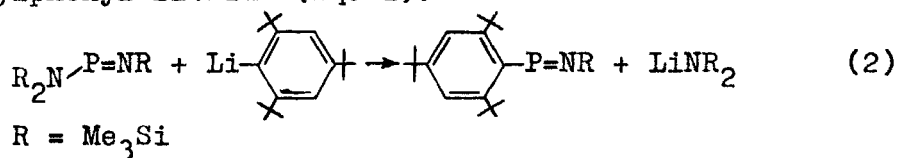
Aminoiminophosphines undergo nucleophilic transamination with the retention of phosphorus coordination when treated with highly hindered lithium amides (Eq.1).<sup>2</sup>



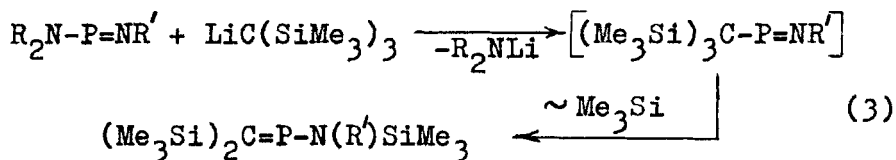
The equilibrium position mainly depends upon the relative nucleophilicity of starting and formed lithium amides and it is displaced toward the less nucleophilic lithium amide.

The reaction described offers a considerable potential for the preparation of new aminoiminophosphi-

nes that are difficult to prepare by classical method. The transfer of this reaction principle to hindered organolithium compounds and aminoiminophosphines provides a pathway to derivatives of dicoordinated phosphorus based on the skeleton C-P=N. The first thermally stable compound of C-P=N type has been synthesised by a nucleophilic displacement reaction between tris(trimethylsilyl)aminoiminophosphine and 2,4,6-tri-*t*-butylphenyl-lithium (Eq. 2).<sup>3,4</sup>

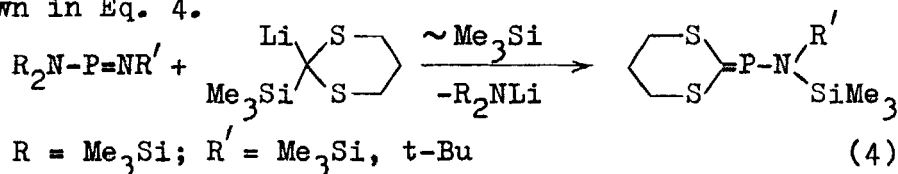


We have observed that aminoiminophosphines react with silyl functionalized organometallic species via nucleophilic displacement at the dicoordinated phosphorus atom with subsequent [1,3]-silyl migration from carbon to nitrogen. For example, treatment of aminoiminophosphines with an equimolar quantity of  $(\text{Me}_3\text{Si})_3\text{CLi}$  in THF at  $-78^\circ\text{C}$  resulted in the formation of aminomethylenephosphines (Eq. 3).



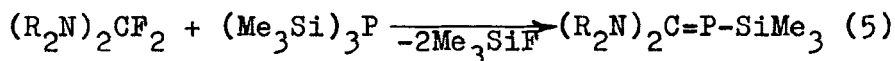
$R = \text{Me}_3\text{Si}; R' = \text{Me}_3\text{Si}, t\text{-Bu}, 1\text{-adamantyl}$

Recently this method has been successfully used by us to prepare aminomethylenephosphines containing thioalkyl-substituents at the methylene carbon atom, as shown in Eq. 4.



P-TRIMETHYLSILYL BIS(DIALKYLAMINO)METHYLENEPHOSPHINES

A number of compounds having the C=P-Si basic structure have been prepared by our group via the condensation of bis(dialkylamino)difluoromethanes with tris(trimethylsilyl)phosphine (Eq. 5).<sup>5</sup>

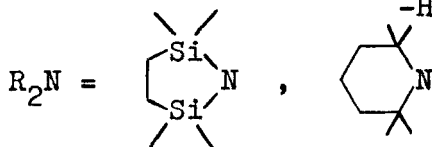
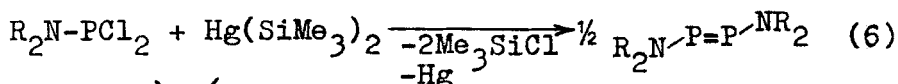


R = Me, Et, i-Pr

A noticeable feature of P-trimethylsilyl bis(dialkylamino)methylenephosphines in comparison with the known methylenephosphines having the C=P-C backbone structure is the upfield shift of the <sup>31</sup>P nmr signal (-27 to -47 ppm). The high reactivity of P-Si bond can be utilized for the synthesis of other types of functionalized dicoordinated phosphorus compounds. Thus, interaction of P-trimethylsilyl methylenephosphines with heterocumulenes (ketenes, isocyanates, isothiocyanates, carbodiimides) leads to compounds based on the skeleton  $\text{>C=P-C=X}$  (X = R<sub>2</sub>C, RN, S).<sup>6</sup> Our investigations in this field are continuing.

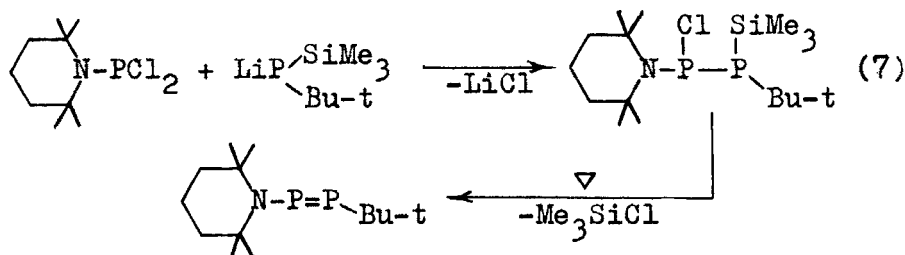
AMINODIPHOSPHENES

On treatment of the sterically hidden aminodichlorophosphines with equimolar quantity of bis(trimethylsilyl)mercury in DME or THF at 20°C the bis(amino)diphosphenes were obtained in quantitative yields.



The composition and constitution of bis(amino)diphosphenes are confirmed by elemental analyses, molecular weight determinations and nmr data. A noteworthy characteristic of the compounds are their P-nmr signals which appear at extremely low field ( $>472$  ppm). The reaction according to Eq. 6 is a general method of converting hindered aminodichlorophosphines into bis-(amino)diphosphenes that can be effected under very mild conditions.

The synthetic approach to the first compound having the N-P=P-C molecular skeleton is outlined in Eq. 7.



The  $^{31}\text{P}\{^1\text{H}\}$  nmr spectrum of 2,2,6,6-tetramethylpiperidino-tert-butyldiphosphene consists of an AB pattern with  $\delta_A +508$ ,  $\delta_B +383$  ppm, and  $^1J(\text{PP})$  611 Hz.

## REFERENCES

1. E. Fluck, Top. Phosphorus Chem., **10**, 193 (1980).
2. V. D. Romanenko, A. V. Ruban, L. N. Markovski, Zh. Obshch. Khim., **53**, 658 (1983).
3. V. D. Romanenko, A. V. Ruban, L. N. Markovski, J. Chem. Soc., Chem. Commun., 187 (1983).
4. V. D. Romanenko, A. V. Ruban, L. N. Markovski, Zh. Obshch. Khim., **53**, 778 (1983).
5. L. N. Markovski, V. D. Romanenko, T. V. Pidvarko, Zh. Obshch. Khim., **52**, 1925 (1982).
6. L. N. Markovski, V. D. Romanenko, T. V. Pidvarko, Zh. Obshch. Khim., **53** (1983).